UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS
International General Certificate of Secondary Education

CHEMISTRY **0620/05**

Paper 5 Practical Test **October/November 2008**

CONFIDENTIAL INSTRUCTIONS

Great care should be taken to ensure that any confidential information given does not reach the candidates either directly or indirectly.

READ THESE INSTRUCTIONS FIRST

The teacher responsible for preparing the examination is **not** allowed to consult the question paper before the examination. Teachers should, as part of the preparation of the examination requirements, carry out any tests indicated on page 2 in order to satisfy themselves that the supplied materials are satisfactory.

The standard Report Form to be included with the scripts is given on pages 7 and 8. Please detach and enclose it with the scripts. If scripts are to be despatched in more than one envelope, it is essential that a copy of the Supervisor's Results and of the Report Form are sent inside **each** envelope.

More material may be issued if required, without penalty, but this should not be necessary. Safety spectacles may be provided if considered necessary.

Supervisors are advised to remind candidates that all substances in the examination should be treated with caution. Please also see under 'General Apparatus' on the use of pipette fillers and safety goggles.

In accordance with COSHH (Control of Substances Hazardous to Health) Regulations, operative in the UK, a hazard appraisal of the examination has been carried out.

Attention is drawn, in particular, to certain materials used in the examination. The following codes are used where relevant.

C = corrosive substance

F = highly flammable substance

H = harmful or irritating substance

O = oxidising substance

T = toxic substance

Hazard data sheets should be available from your suppliers.

If you have any queries regarding these Instructions, please contact CIE
by e-mail: International@cie.org.uk,
by phone: +44 1223 553554,
by fax: +44 1223 553558,
stating the Centre number, the nature of the query and the syllabus number quoted above.

This document consists of **5** printed pages and **3** blank pages.

For Question 1

Each candidate will need the following

(a) a 50 cm^3 measuring cylinder;

(b) rack of test-tubes;

[H] **(c)** For Experiment 1, 4 g of white anhydrous copper(II) sulphate,$CuSO_4$, in a stoppered container, labelled solid **A**;

[H] **(d)** For Experiment 2, 4 g of blue hydrated copper(II) sulphate crystals, $CuSO_4.5H_2O$, in a stoppered container, labelled solid **B**;

(e) For Experiment 3, 4 g of ammonium chloride in a stoppered container, labelled solid **C**;

(f) For Experiment 4, 4 g of sodium hydrogencarbonate, $NaHCO_3$, in a stoppered container, labelled solid **D**;

(g) Four polystyrene cups;

(h) One stirring thermometer, covering the range 0 °C to 100 °C;

(i) Distilled water;

(j) Teat pipette;

(k) Stop clock or access to a timer.

A 4 g sample of ammonium chloride when added to 30 cm^3 of distilled water should give a temperature drop of about 10 °C.

For Question 2

Each candidate will require

[H] (a) 20 cm^3 of an aqueous solution of copper(II) chloride, $CuCl_2.2H_2O$, of concentration 0.25 mol / dm^3, labelled solution **K**;

[C] (b) 20 cm^3 of an aqueous solution of iron(III) chloride, $FeCl_3$, of concentration 0.25 mol / dm^3, labelled solution **L**;

(c) Universal indicator paper and pH chart;

[C] (d) aqueous sodium hydroxide of concentration 2 mol / dm^3, labelled **aqueous sodium hydroxide**;

[C] (e) aqueous ammonia solution of concentration 2 mol / dm^3, labelled **aqueous ammonia**;

[T] (f) barium chloride solution of sufficient concentration to give a positive sulphate test, labelled **barium chloride solution**;

[C] (g) aqueous hydrochloric acid of concentration 2 mol / dm^3, labelled **hydrochloric acid**;

[C] (h) aqueous nitric acid of concentration 1 mol / dm^3, labelled **nitric acid**;

[C] (i) aqueous silver nitrate solution of sufficient concentration to give a positive chloride test, labelled **silver nitrate solution**;

(j) rack of test-tubes;

(k) distilled water;

(l) teat pipettes;

(m) one measuring cylinder, 10 cm^3.

THE SUPERVISOR'S REPORT IS ON PAGES 7 & 8

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This form must be completed and returned in the envelope with the scripts.

REPORT ON PRACTICAL CHEMISTRY

NOVEMBER 2008

1 **(a)** Supervisor's Results

It is recommended that the supervisor should be a chemistry teacher.

The supervisor is asked to carry out the experiments in Questions 1 and 2 and to record the results on a spare copy of the question paper clearly labelled 'Supervisor's Results'. Failure to enclose these results and this report form may lead to candidates being unavoidably penalised.

(b) The Candidate Numbers of candidates in each session were:

First Session	*Second Session*

2 The Supervisor is invited to report details of any difficulties experienced by candidates, giving names and Candidate Numbers. The report should include reference to:

(a) any general difficulties encountered in making preparations for the examination;

(b) difficulties due to faulty apparatus or materials;

(c) accidents to apparatus or materials.

Other cases of individual hardship, e.g. illness, temporary disability, should be reported direct to UCLES on the normal *Application for Special Consideration* form.

NAME OF CENTRE .....

CENTRE NUMBER .....

SIGNED .....
Supervisor

DECLARATION (to be signed by the Principal)

The preparation of this practical examination has been carried out as to maintain fully the security of the examination.

NAME .....
(in block capitals)

SIGNED ..... (Principal)